### भारतीय मानक Indian Standard

## इस्पात के रासायनिक विश्लेषण की पद्धति

IS 228 (Part 4): 2024

भाग 4 ग्रेविमेट्रिक विधि द्वारा कुल कार्बन का निर्धारण (कार्बन ≥ 0.1 प्रतिशत के लिए <mark>ड्</mark>डि

( चौथा पुनरीक्षण )

# Method for Chemical Analysis of Steels

Part 4 Determination of Total Carbon by Gravimetric Method (for Carbon ≥ 0.1 Percent)

( Fourth Revision )

ICS 77.080.20

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भारतीय मानक ब्यूरो

BUREAU OF INDIAN STANDARDS मानक भवन, 9 बहादुर शाह ज़फर मार्ग, नई दिल्ली - 110002 MANAK BHAVAN, 9 BAHADUR SHAH ZAFAR MARG NEW DELHI - 110002

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#### **FOREWORD**

This Indian Standard (Part 4) (Fourth Revision) was adopted by the Bureau of Indian Standards after the draft finalized by the Methods of Chemical Analysis of Metals Sectional Committee had been approved by the Metallurgical Engineering Division Council.

This standard was first published in 1952 and subsequently revised in 1959, 1974 and 1987, covered the chemical analysis of plain carbon and low alloy steels, along with pig iron and cast iron. It was revised again to make it comprehensive in respect of steel analysis and to exclude pig iron and cast iron which were being covered in separate standards. During its second revision the standard has been split up in several parts.

This revision has been brought out to bring the standard in the latest style and format of the Indian Standards.

This part covers method for determination of total carbon content of plain carbon, low alloy and high alloy steels of 0.1 percent and above by the gravimetric method. The other parts of this series are:

Part 1	Determination of carbon by volumetric method (for carbon 0.05 to 2.50 percent)			
Part 2	Determination of manganese in plain-carbon and low alloy steels by arsenite method			
Part 3	Determination of phosphorus by alkalimetric method			
Part 5	Determination of nickel by dimethyl glyoxime (gravimetric) method (for nickel $\geq 0.1$ percent)			
Part 6	Determination of chromium by persulphate oxidation method (for chromium $\geq 0.1$ percent)			
Part 7	Determination of molybdenum by alpha-benzoinoxime method (for molybdenum $> 1$ percent and not containing tungsten)			
Part 8	Determination of silicon by gravimetric method (for silicon 0.05 to 5.00 percent)			
Part 9	Determination of sulphur by evolution method (for sulphur 0.01 to 0.25 percent)			
Part 10	Determination of molybdenum by thiocyanate (photometric) method in low and high alloy steels (for molybdenum 0.01 to 1.5 percent)			
Part 11	Determination of total silicon by reduced molybdosilicate spectrophotometric method in carbon steels and low alloy steels (for silicon 0.01 to 0.05 percent)			
Part 12	Determination of manganese by periodate spectrophotometric method in plain carbon, low alloy and high alloy steels (for manganese 0.01 to 5.0 percent)			
Part 13	Determination of arsenic			
Part 14	Determination of carbon by thermal conductivity method (for carbon 0.005 to 2.000 percent)			
Part 15	Determination of copper by thiosulphate iodide method (for copper 0.05 to 5 percent)			
Part 16	Determination of tungsten by spectrophotometric method (for tungsten 0.1 to 2 percent)			
Part 17	Determination of nitrogen by thermal conductivity method (for nitrogen up to 0.04 percent)			
Part 18	Determination of oxygen by instrumental method (for oxygen 0.01 to 0.100 0 percent)			
Part 19	Determination of nitrogen by steam distillation method (for nitrogen 0.002 to 0.50 percent)			
Part 20	Determination of total carbon and sulfur content — Infrared absorption method after combustion in an induction furnace (routine method)			
Part 21	Determination of copper by spectrometric method (for copper 0.02 to 0.5 percent)			
Part 22	Determination of total hydrogen in steel by thermal conductivity method (hydrogen $0.1~\mathrm{ppm}$ to $50~\mathrm{ppm}$ )			

#### Indian Standard

#### METHOD FOR CHEMICAL ANALYSIS OF STEELS

## PART 4 DETERMINATION OF TOTAL CARBON BY GRAVIMETRIC METHOD (FOR CARBON ≥ 0.1 PERCENT)

( Fourth Revision )

#### 1 SCOPE

This standard (Part 4) covers the method for determination of total carbon content of plain carbon, low alloy and high alloy steels of 0.1 percent and above by the gravimetric method.

#### 2 REFERENCES

This standard given below contain provisions which through reference in this text, constitute provisions of this standard. At the time of publication, the editions indicated was valid. All standards are subject to revision and parties to agreements based on this standard are encouraged to investigate the possibility of applying the most recent edition of this standard:

IS No Title

IS 6226 (Part 1): Recommendations of apparatus for chemicals analysis of metals: Part 1 Apparatus for determination of carbon by direct combustion (first revision)

#### 3 OUTLINE OF THE METHOD

The sample is burnt in a stream of purified oxygen and the carbon dioxide formed is absorbed, after purification, in suitable absorbent and determined.

#### **4 REAGENTS**

- 4.1 Oxygen (O<sub>2</sub>) 99.5 percent minimum
- 4.2 Accarite or Soda Lime 0.80 mm to 2.00 mm
- **4.3 Magnesium Perchlorate** Mg (ClO<sub>4</sub>)<sub>2</sub>, 0.80 mm to 2.00 mm

#### 4.4 Boat/Crucible

Boat/crucible of precise dimension for accommodating in the resistance and induction furnace.

Pre-ignite the boats/crucibles in air or oxygen in a furnace for an hour at 1 100 °C and store in a desiccator and check for consistency of the blank values.

**4.5 Flux/Accelerator** — low carbon copper, red lead (pre-ignited at 550 °C), tin and iron of low carbon content.

#### **5 APPARATUS**

- **5.1** The apparatus recommended in IS 6226 (Part l) may be used.
- **5.2** Instead of the resistance furnace, an induction furnace may also be used.

#### 6 SAMPLING

- **6.1** The sample shall be drawn as prescribed in the relevant Indian Standards.
- **6.2** The sample is to be cleaned with analytical grade ether and acetone, dried in an air oven at  $100~^{\circ}\text{C} \pm 5~^{\circ}\text{C}$ .

#### 7 PROCEDURE

**7.1** Assemble the apparatus. Switch on the furnace, if it is a resistance furnace, and allow it to attain a temperature of 1 050 °C (*see* Note), all the while passing oxygen through the apparatus so that it bubbles freely at the exit end of the train. Disconnect the absorption bulb, keep in a desiccator till it attains room temperature and take the initial weight. Repeat the operation till a constant weight is obtained.

NOTE — For high chromium and high nickel steel, the temperature of 1 250  $^{\circ}$ C is recommended for complete combustion.

**7.2** Weigh to the nearest 0.001 g, 2.0 g to 3.0 g of the test sample. Transfer to the pre-ignited combustion boat covered at the bottom with a thin layer of calcined alumina. Spread the sample evenly over the top of the alumina and cover it with 2.0 g to 3.0 g of the flux. Introduce the boat slowly in the hot zone of the combustion tube.

To access Indian Standards click on the link below:

- **7.3** In the case of induction heating, weigh to the nearest 0.001 g, 0.9 g to 1.1 g of the sample and transfer to a pre-ignited crucible. Add an equal quantity of the flux. Place the crucible in position on the pedestal post of the furnace, raise to the combustion position and lock the system. Pass oxygen through the system and ignite the sample.
- **7.4** Maintain a rapid flow of oxygen (800 ml/min to 1 000 ml/min) throughout the combustion, then reduce to 400 ml to 500 ml per minute and maintain it for another 6 min to 8 min in order to sweep out the carbon dioxide.
- **7.5** Remove the absorption bulb and weigh it after keeping it in desiccator till it attains room temperature. The increase in weight of the bulb represents the carbon dioxide.
- **7.6** Remove the boat or crucible and examine for any incomplete combustion. If the sample is not thoroughly fused, repeat the determination with a fresh sample.

#### 7.7 Blank

Charge a pre-ignited boat or crucible, as the case may be, with the same amount of flux used in the determination and follow the procedure as in  $\frac{7.2}{1.5}$  to  $\frac{7.5}{1.5}$ .

#### **8 CALCULATION**

**8.1** Calculate the total carbon content of the sample as follows:

Carbon, percent = 
$$\frac{A-B}{C} \times 27.29$$

where

- A = increase in mass, in g, of the absorption bulb due to carbon dioxide from the sample;
- *B* = increase in mass, in g, of the absorption bulb due to carbon dioxide from the blank determination; and
- C = mass, in g, of the sample taken.

#### 9 ACCURACY

The accuracy of the method is  $\pm$  0.01 percent for carbon the range of 0.1 percent to 0.75 percent and  $\pm$  0.02 percent for carbon above 0.75 percent.

#### ANNEX A

(Foreword)

#### **COMMITTEE COMPOSITION**

Methods of Chemical Analysis of Metals Sectional Committee, MTD 34

Organization Representative(s)

CSIR - National Metallurgical Laboratory, Jamshedpur DR SANCHITA CHAKRAVARTY (*Chairperson*)

Arcelor Mittal Nippon Steel, Mumbai Shri Manoj Gupta

SHRI KIRIT TAILOR (Alternate)

Bhabha Atomic Research Centre, Mumbai Ms Sanjukta A. Kumar

SHRI M. V. RANA (Alternate)

CSIR - National Metallurgical Laboratory, Jamshedpur DR ASHOK K. MOHANTY (Alternate)

Defence Metallurgical Research Laboratory, Ministry of SHRI S. S. KALYAN KAMAL

Defence, Hyderabad

Directorate General of Quality Assurance, Ministry of SHRI A. MITRA

Defence, New Delhi

SHRI D. KARTIKEY (Alternate)

Geological Survey of India, New Delhi Shri Nitin Purushottam

SHRIMATI SANJUKTA DEY PAL (Alternate)

Hindalco Industries Limited, Mumbai Shri Krishanu Mahapatra

SHRI ASHUTOSH ACHARYA (Alternate)

Indian Metals and Ferro Alloys Limited, Bhubaneswar Shri Dinesh Kumar Mohanty

Jawaharlal Nehru Aluminium Research Development and

Design Centre, Nagpur

DR UPENDRA SINGH

JSW Steel Limited, Mumbai Shri Kotrabasavaraju

SHRI MARULASIDDESHA U. M. (Alternate)

National Aluminium Company Limited, Bhubaneswar Shrimati Sukla Nandi

SHRI DEBANANDA BHATTACHARYYA (Alternate)

National Test House, Kolkata

DR RAJEEV KUMAR UPADHYAY

SHRI AKBAR H. (Alternate)

Shriram Institute for Industrial Research, Delhi DR LAXMI RAWAT

SHRI PUNEET KAPOOR (Alternate)

Steel Authority of India Limited - Salem Steel Plant, SHRI L. SIVAKUMAR

Salem

SHRI VIVEKANANDHAN G. (Alternate)

Tata Steel Limited, Kolkata Shri Dr Jatin Mohapatra

DR RAVIKRISHNA CHATTI (Alternate)

BIS Directorate General Shri Sanjiv Maini, Scientist 'F'/Senior Director

AND HEAD (METALLURGICAL ENGINEERING) [REPRESENTING DIRECTOR GENERAL (Ex-officio)]

Member Secretary
SHRI ASHISH PRABHAKAR WAKLE
SCIENTIST 'C'/DEPUTY DIRECTOR
(METALLURGICAL ENGINEERING), BIS

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#### (Continued from second cover)

- Part 23 Determination of total nitrogen in steel by optical emission spectrometer (nitrogen 0.002 to 1.0 percent)
- Part 24 Determination of nitrogen in steel by inert gas fusion Thermal conductivity method (nitrogen 0.001 to 0.2 percent)

The composition of the Committee responsible for formulation of this standard is given in Annex A.

In reporting the result of a test or analysis made in accordance with this standard, is to be rounded off, it shall be done in accordance with IS 2: 2022 'Rules for rounding off numerical values (*second revision*)'. The number of significant places retained in the rounded off value should be same as that of the specified value in this standard.

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#### **Amendments Issued Since Publication**

Amend No.	Date of Issue	Text Affected	

#### **BUREAU OF INDIAN STANDARDS**

#### **Headquarters:**

Manak Bhavan, 9 Bahadur Shah Zafar Marg, New Delhi 110002

Telephones: 2323 0131, 2323 3375, 2323 9402 Website: www.bis.gov.in

Regional Offices:		
Central : 601/A, Konnectus Tower -1, 6 <sup>th</sup> Floor, DMRC Building, Bhavbhuti Marg, New Delhi 110002	{ 2323 7617	
Eastern : 8 <sup>th</sup> Floor, Plot No 7/7 & 7/8, CP Block, Sector V, Salt Lake, Kolkata, West Bengal 700091	{ 2367 0012 2320 9474 { 265 9930	
Northern: Plot No. 4-A, Sector 27-B, Madhya Marg, Chandigarh 160019	{ 265 9930	
Southern: C.I.T. Campus, IV Cross Road, Taramani, Chennai 600113	2254 1442 2254 1216	
Western: Manakalya, 5 <sup>th</sup> Floor/MTNL CETTM, Technology Street, Hiranandani Gardens, Pow Mumbai 400076	ai { 25700030 25702715	

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